

Synthetic Studies toward Potent Cytotoxic Agents Amphidinolides: Synthesis of the C₁-C₁₈ Moiety of Amphidinolides G, H and L

T. K. Chakraborty* and V. R. Suresh

Indian Institute of Chemical Technology, Hyderabad 500 007, India

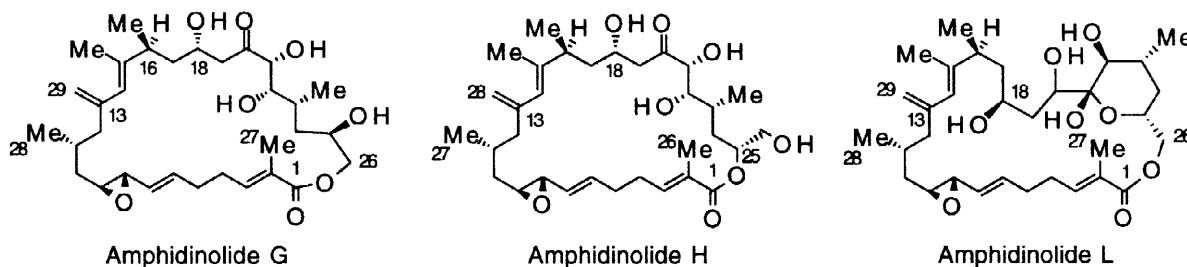
Received 3 September 1998; accepted 21 September 1998

Abstract

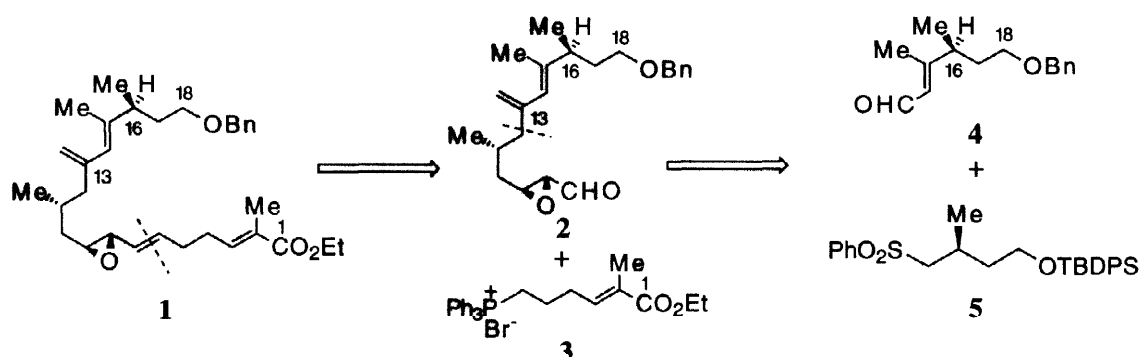
Stereoselective synthesis of the (8*S*, 9*S*, 11*R*, 16*S*)-C₁-C₁₈ segment **1** of amphidinolides G, H and L, bearing the unique trisubstituted “*s-cis*-1,3-diene” moiety (C₂₈₍₂₉₎=C₁₃-C₁₄=C₁₅), has been achieved for the first time following a highly efficient convergent strategy. © 1998 Elsevier Science Ltd. All rights reserved.

Keywords: amphidinolides; antitumor compounds; macrolides; asymmetric alkylation.

The amphidinolides constitute a family of structurally complex macrolide molecules isolated from marine sources. Many of them have potent toxicity against various tumor cell lines [1,2]. Total synthesis of none of these compounds has so far been reported [3-15]. One of the major obstacles encountered en route to the total synthesis of some of the important members of this family, like amphidinolides B, D, G, H and L, is the construction of an uncommon trisubstituted “*s-cis*-1,3-diene” moiety (C₂₈₍₂₉₎=C₁₃-C₁₄=C₁₅), present in these molecules, in its naturally occurring configuration. We report here the first synthesis of this very important structural entity followed by its elaboration to the (8*S*, 9*S*, 11*R*, 16*S*)-C₁-C₁₈ fragment of amphidinolides G, H and L.

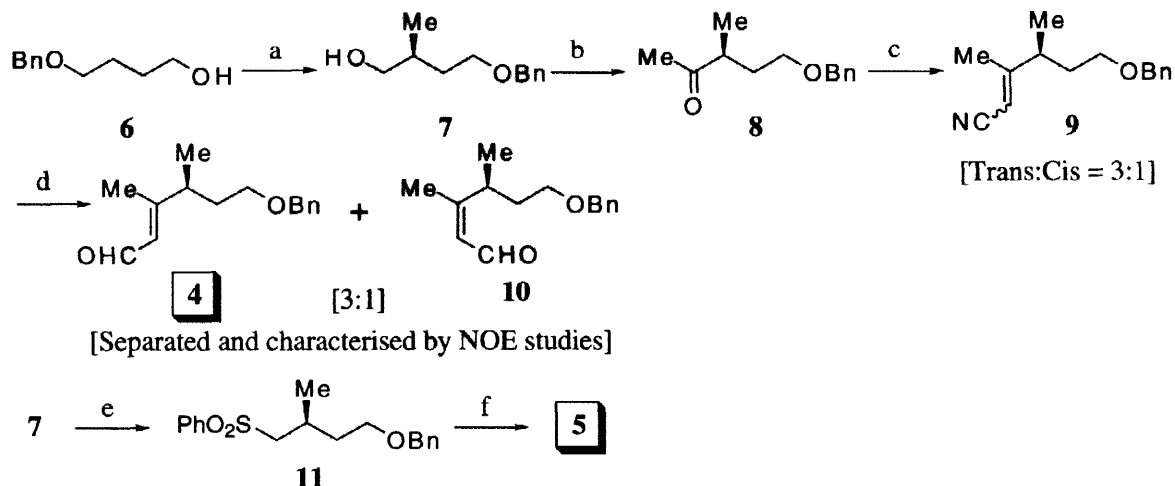


Retrosynthetically, the C₁-C₁₈ fragment **1** can be divided, as shown in Scheme 1, into two halves: the C₇-C₁₈ fragment **2** bearing the diene and the epoxyaldehyde functionalities and the C₁-C₆ Wittig component **3**. The important diene moiety of **2** was planned to be constructed by coupling the *E*- α,β -unsaturated aldehyde **4** and the functionalized sulfone unit **5**.



Scheme 1. Retrosynthetic analysis of **1**.

Scheme 2 outlines the syntheses of fragments **4** and **5**. Use of a common chiral precursor, (*S*)-4-benzyloxy-2-methylbutan-1-ol (**7**), for the syntheses of both the fragments is the salient feature of this scheme. The monobenzyl-protected butane-1,4-diol was transformed into **7** by Evans asymmetric alkylation method following reported procedures [16–18]. The chiral alcohol **7** was then converted to the methylketo intermediate **8** in 3 steps in 75% overall yield. Horner-Wadsworth-Emmons olefination of **8** with diethyl cyanomethylphosphonate gave a mixture of acrylonitriles **9** (3:1) in 98% yield. Reduction of this mixture of nitriles with DIBAL afforded the isomeric aldehydes **4** and **10** (3:1), in 93% yield, which could be separated easily at this stage by silica gel column chromatography. That the major isomer was the required *E*-olefin **4** was confirmed by ¹H NOE difference spectroscopic studies. Irradiation of the olefinic C₁₄-H signal of **10** at δ 5.87 caused significant enhancement of the 29-Me resonance at δ 1.87, indicating a *cis*-relationship between them. As expected, there was no NOE observed between the C₁₄-H and the 29-Me in the *E*-olefin **4**.

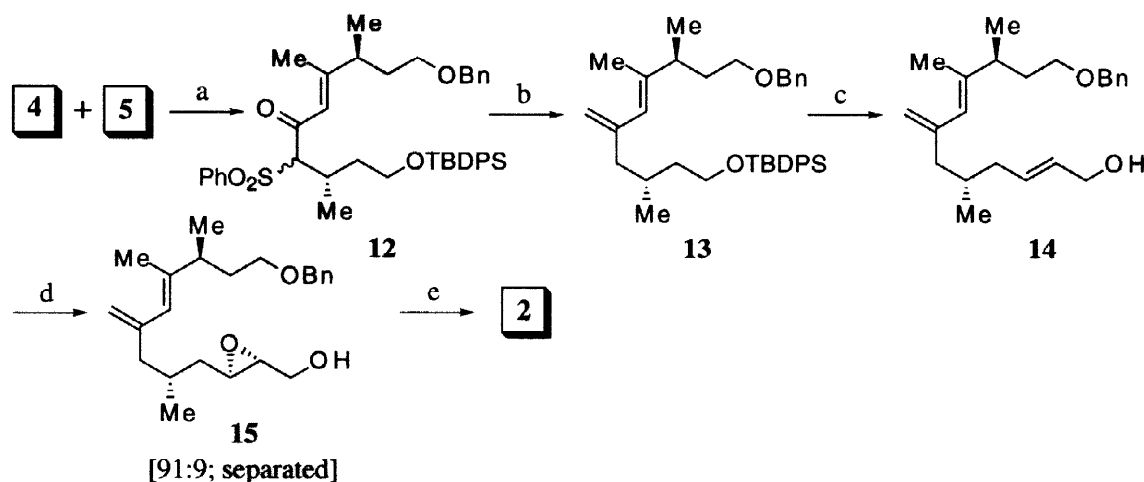


Scheme 2. Reagents and conditions. a) Ref. 16–18; b) (i) (COCl)₂ (1.5 eq.), DMSO (3.2 eq.), Et₃N (5 eq.), CH₂Cl₂, -78 to 0 °C, 1.5 h; (ii) MeMgI (2.0 M, 2 eq.), Et₂O, 0 to 25 °C, 1 h; (iii) same as step (i), 75% from **7**; c) (Et₂O)₂P(O)CH₂CN (1.5 eq.), NaH (1.5 eq.), DME, 25 °C, 1 h, then **8** in DME, 25 °C, 3 h, 98%; d) DIBAL (1.1 eq.), toluene, -78 °C, 1 h, 93%; e) (i) MsCl (1.2 eq.), pyridine, 25 °C, 3 h; (ii) PhSH (1.2 eq.), K₂CO₃ (1.5 eq.), DMF, 25 °C, 0.5 h; (iii) *m*CPBA (4 eq.), CH₂Cl₂, 0 to 25 °C, 3 h; 89% from **7**; f) (i) H₂, Pd/C, MeOH, 25 °C, 1 h; (ii) TBDPSCl (1.2 eq.), Et₃N (2 eq.), DMAP (0.1 eq.), CH₂Cl₂, 25 °C, 3 h; 95% from **11**.

For the synthesis of **5**, the common chiral precursor **7** (Scheme 2) was converted to the phenylsulfone **11** in 3 steps in 89% overall yield. This was followed by a change in the protective group, which was necessary to differentiate the two protective groups at a later stage, furnishing the requisite TBDPS-protected sulfone **5**.

Coupling of fragments **4** and **5** and further elaboration of the resulting coupled product to the advanced stage intermediate **2** is delineated in Scheme 3. Addition of the anion generated from **5** to the aldehyde **4** gave a diastereomeric mixture of β -hydroxysulfones which were oxidized using *O*-iodoxybenzoic acid (IBX) [19] to ketones **12** in 80% overall yield. Removal of the phenylsulfone appendage using lithium naphthalenide (LN) (60% yield) [20] and subsequent one-carbon Wittig olefination furnished the intermediate **13**^{1,2} in 88% yield, thus, completing successfully the first synthesis of the targeted “*s-cis*-1,3-diene” moiety.

Routine functional group manipulations converted **13** to the allylic alcohol **14** in 4 steps in 80% overall yield. Sharpless asymmetric epoxidation [21] of **14** with natural (+)-diethyl L-tartrate gave the expected (8*S*,9*S*)-epoxy alcohol **15** as the major product (in 91:9 ratio). The minor diastereomer could be easily separated by silica gel column chromatography. The epoxyalcohol **15** was subsequently oxidized to get the intermediate **2**.



Scheme 3. Reagents and conditions. a) (i) **5** (1 eq.), ^tBuLi (1 eq.), THF, -78 °C, 20 min., then **4** in THF, -78 to 0 °C, 1 h; (ii) IBX (2 eq.), DMSO, 25 °C, 1 h, 80% in 2 steps; (b) (i) LN (excess), THF, -78 °C, 1 h, 60%; (ii) Ph₃P=CH₂ (2 eq.), Et₂O, 0 °C, 0.5 h, 88%; (c) (i) TBAF (1.5 eq.), THF, 25 °C, 5 h; (ii) (COCl)₂ (1.5 eq.), DMSO (3.2 eq.), Et₃N (5 eq.), CH₂Cl₂, -78 to 0 °C, 1.5 h; (iii) Ph₃P=CHCO₂Et (2 eq.), C₆H₆, 25 °C, 1 h; (iv) DIBAL (2.2 eq.), CH₂Cl₂, -78 °C, 1 h, 80% from **13**; d) Ti(ⁱPrO)₄ (0.2 eq.), (+)-DET (0.22 eq.), TBHP (2 eq.), CH₂Cl₂, -10 °C, 12 h, 92% (based on 40% recovered starting material); e) same as step c(ii).

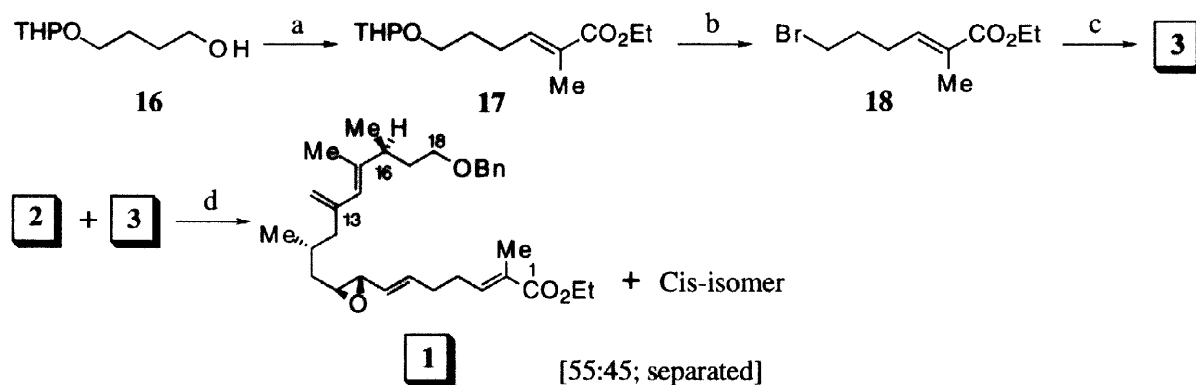
Scheme 4 describes the preparation of the phosphonium salt **3** and its use in the olefination of epoxy aldehyde **2**. Mono-THP-protected butane-1,4-diol **16** was oxidized and olefination with stabilized ylide gave the α,β -unsaturated ester **17**. Deprotection of the THP-ether, bromination of the hydroxyl group and finally, treatment with Ph₃P gave the Wittig salt **3**.

Finally, the ylide generated from **3** was reacted with the aldehyde **2**, following the procedure reported by Kobayashi *et al* [13], giving a mixture of olefins (*trans*:*cis* = 55:45) which were separated by preparative TLC to furnish the desired C₁-C₁₈ fragment **1**.^{1,3}

¹ Satisfactory NMR, IR and mass spectra were obtained for this compound.

² **13**: [α]_D²² = -10.4 (c 0.5, CHCl₃); ¹H NMR (CDCl₃, 200 MHz): δ 7.7-7.2 (m, 15 H, aromatic), 5.54 (s, 1 H, C₁₄-H), 4.91 and 4.75 (two s, 2 H, C=CH₂), 4.42 (s, 2 H, CH₂Ph), 3.67 (t, *J* = 6.8 Hz, 2 H, CH₂OTBDPS), 3.35 (t, *J* = 6.8 Hz, 2 H, CH₂OBn), 2.32 (m, 1 H, C₁₆-H), 2.06 (dd, *J* = 14.8, 5.6 Hz, 1 H, C₁₂-H), 1.88-1.52 (m, 4 H, C₁-H, C₁₂-H, C₁₇-H₂), 1.68 (s, 3 H, C₁₅-CH₃), 1.3 (m, 2 H, C₁₀-H₂), 1.04 (s, 9 H, SiPh₂Bu), 1.0 (d, *J* = 6.7 Hz, 3 H, C₁₆-CH₃), 0.78 (d, *J* = 6.3 Hz, 3 H, C₁₁-CH₃).

³ **1**: [α]_D²² = 14.5 (c 0.2, CHCl₃); ¹H NMR (CDCl₃, 400 MHz): δ 7.32 (m, 5 H, aromatic), 6.74 (t, *J* = 7 Hz, 1 H, C₉-H), 5.69 (dt, *J* = 15.2, 7.7 Hz, 1 H, C₆-H), 5.57 (s, 1 H, C₁₄-H), 5.08 (dd, *J* = 15.2, 8.1 Hz, 1 H, C₇-H), 4.96 and 4.81 (two s, 2 H, C₁₃=CH₂), 4.48 (s, 2 H, CH₂Ph), 4.19 (q, *J* = 7 Hz, 2 H, CO₂CH₂CH₃), 3.4 (t, *J* = 7 Hz, 2 H, CH₂OBn), 3.29 (dd, *J* = 8.1, 2.1 Hz, 1 H, C₈-H), 2.8 (dt, *J* = 7, 2.1 Hz, 1 H, C₉-H), 2.45-1.88 (m, 8 H, allylic, C₁₁-H, C₁₆-H), 1.84 (s, 3 H, C₂-CH₃), 1.8-1.58 (m, 4 H, C₁₀-H₂, C₁₇-H₂), 1.7 (s, 3 H, C₁₅-CH₃), 1.28 (t, *J* = 7 Hz, CO₂CH₂CH₃), 1.03 (d, *J* = 6.8 Hz, 3 H, C₁₆-CH₃), 0.9 (d, *J* = 6.7 Hz, C₁₁-CH₃).



Scheme 4. Reagents and conditions. a) (i) $(\text{COCl})_2$ (1.5 eq.), DMSO (3.2 eq.), Et_3N (5 eq.), CH_2Cl_2 , -78 to 0 °C, 1.5 h; (ii) $\text{Ph}_3\text{P}=\text{C}(\text{CH}_3)\text{CO}_2\text{Et}$ (2 eq.), C_6H_6 , 25 °C, 1 h, 90% from **16**; b) (i) PTSA (cat.), MeOH, 25 °C, 2 h; (ii) CBr_4 (2.5 eq.), Ph_3P (2.5 eq.), CH_2Cl_2 , 0 to 25 °C, 2 h, 92% from **17**; c) Ph_3P (1.2 eq.), CH_3CN , reflux, 12 h, 85%; d) **3** (2 eq.), $^t\text{BuLi}$ (2 eq.), THF, -78 °C, 2 h, then **2** in THF, -78 to 25 °C, 12 h, 80%.

In conclusion, an efficient convergent route presented here led to the first stereoselective synthesis of the (8*S*, 9*S*, 11*R*, 16*S*)- C_1 - C_{18} segment of amphidinolides G, H, and L which will help to achieve the total synthesis of these molecules. Further work is under progress.

Acknowledgements

We thank Drs. A. C. Kunwar and M. Vairamani for NMR and mass spectroscopic assistance, respectively; CSIR, New Delhi for research fellowship (V. R. S.) and Young Scientist Award Research Grant (T.K.C.).

References

- [1] Ishibashi M, Kobayashi J. *Heterocycles* 1997;44:543-572.
- [2] Kobayashi J, Ishibashi M. *Chem. Rev.* 1993;93:1753-1789.
- [3] For previous synthetic studies toward various fragments of amphidinolides see references 4-15.
- [4] O'Connor SJ, Williard PG. *Tetrahedron Lett.* 1989;30:4637-4640.
- [5] Boden C, Pattenden G. *Synlett* 1994;181-182.
- [6] Tsuda M, Sasaki T, Kobayashi J. *J. Org. Chem.* 1994;59:3734-3737.
- [7] Ishibashi M, Ishiyama H, Kobayashi J. *Tetrahedron Lett.* 1994;35:8241-8242.
- [8] Eng HM, Myles DC. Abstracts of the papers of the American Chemical Society 1995;209:No. Pt 2, 405 ORGN1.
- [9] Chakraborty TK, Thippeswamy D, Suresh VR, Jayaprakash S. *Chemistry Lett.* 1997;563-564.
- [10] Chakraborty TK, Suresh VR. *Chemistry Lett.* 1997;565-566.
- [11] Lee D-H, Lee S-W. *Tetrahedron Lett.* 1997;38:7909-7910.
- [12] Tsuda M, Hatakeyama A, Kobayashi J. *J. Chem. Soc. Perkin Trans. 1* 1998;149-155.
- [13] Kobayashi J, Hatakeyama A, Tsuda M. *Tetrahedron* 1998;54:697-704.
- [14] Hollingworth GJ, Pattenden G. *Tetrahedron Lett.* 1998;39:703-706.
- [15] Cid MB, Pattenden G. *Synlett* 1998;540-542.
- [16] Evans DA, Ennis MD, Mathre DJ. *J. Am. Chem. Soc.* 1982;104:1737-1738.
- [17] Smith III AB, Hale KJ. *Tetrahedron Lett.* 1989;30:1037-1040.
- [18] Boeckman Jr. RK, Barta TE, Nelson SG. *Tetrahedron Lett.* 1991;32:4091-4094.
- [19] Frigerio M, Santagostino M. *Tetrahedron Lett.* 1994;35:8019-8022.
- [20] Jones AB, Villalobos A, Linde II RG, Danishefsky SJ. *J. Org. Chem.* 1990;55:2786-2797.
- [21] Gao Y, Hanson RM, Klunder JM, Ko SY, Masamune H, Sharpless KB. *J. Am. Chem. Soc.* 1987;109:5765-5780.